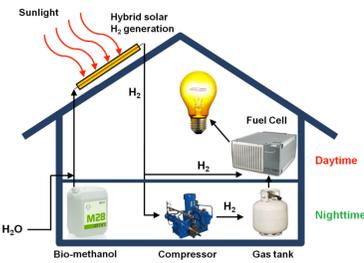


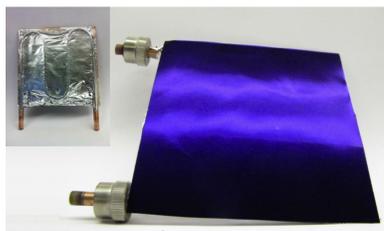
Introduction

- Tremendous research efforts in capturing thermal energy from the sun
- 2 main areas: non-concentrated vs. concentrated solar power (CSP)
- Non-concentrated solar power has limits to temperatures and therefore applications
- Current application: steam reforming of biofuels into hydrogen
- Denser biofuels require higher outlet temperatures, meaning CSP is necessary

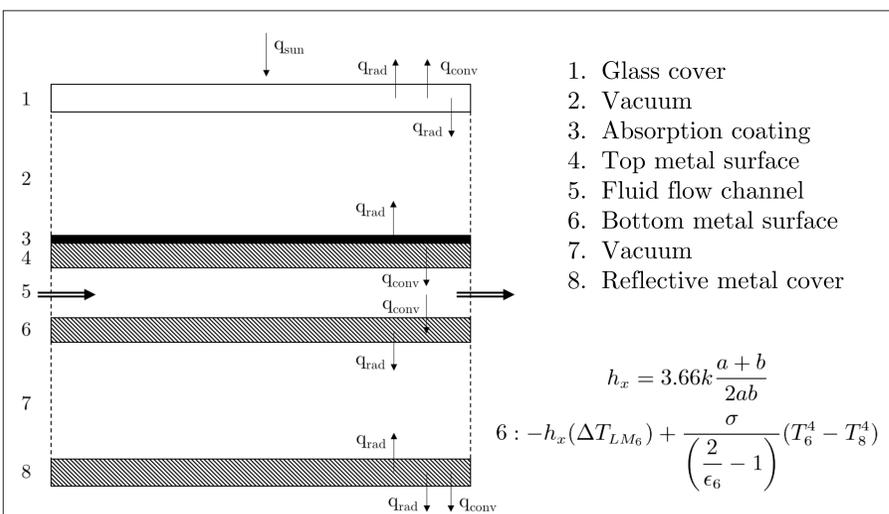


Objectives

- Goal: develop a numerical heat transfer model to predict collector temperatures as a function of concentration
- Model based on existing novel non-concentrating solar thermal collector
- Experimental setup consists of flat plate collector inside an insulating vacuum
- Predict temperatures of all collector components to determine efficiency

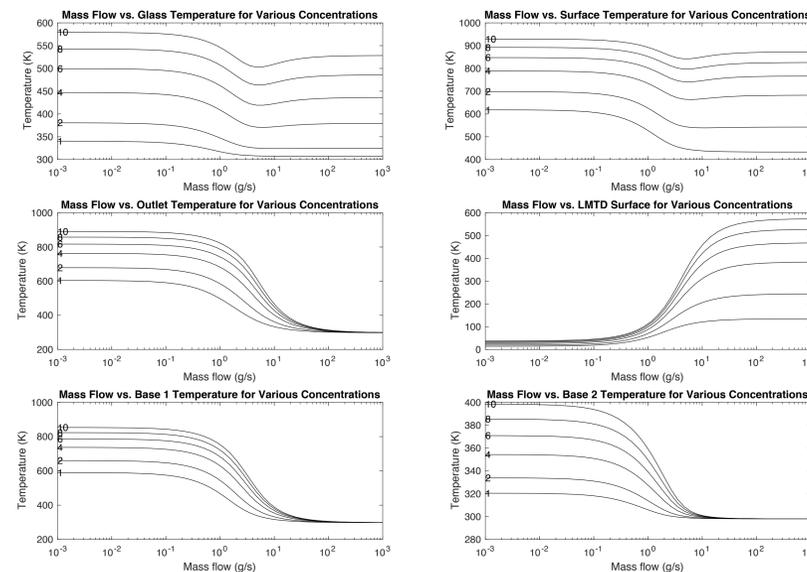


Methods

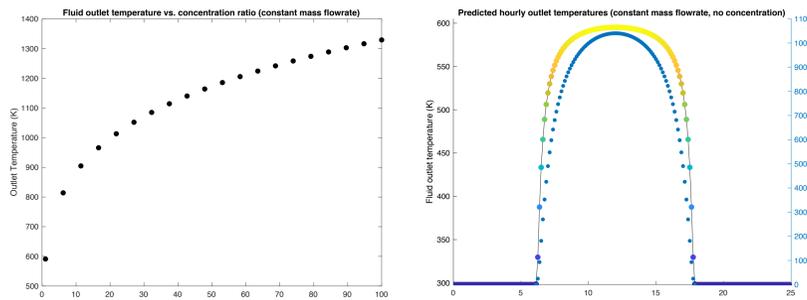


- The above diagram was used to derive the governing heat transfer equations
- Most surfaces had simple radiative/convective energy balances
 - Example of energy balance for surface 6 shown above
 - Convection from fluid to surface 6, radiation from surface 6 to surface 8
 - Required derivation of ΔT_{LM} for top and bottom surfaces
- Imported coupled equations into Matlab to numerically solve for temperatures as a function of flowrate, vary q_{sun} to determine effects of concentration

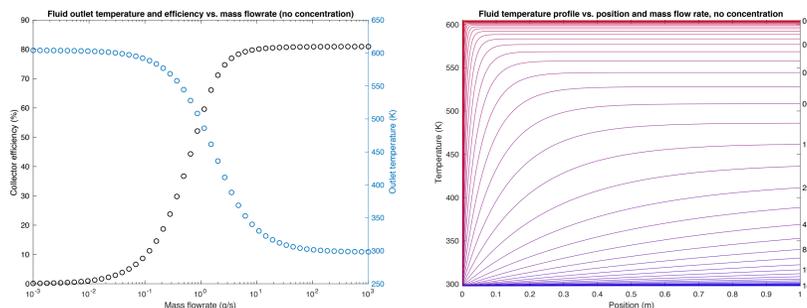
Results



- Predicted temperatures for surfaces 1, 3, 6, 8, and fluid outlet temperature
- Logarithmically scaled mass flow rates, concentrations ranging from 1 – 10
- Can see stagnation temperature and how temperature changes with flow



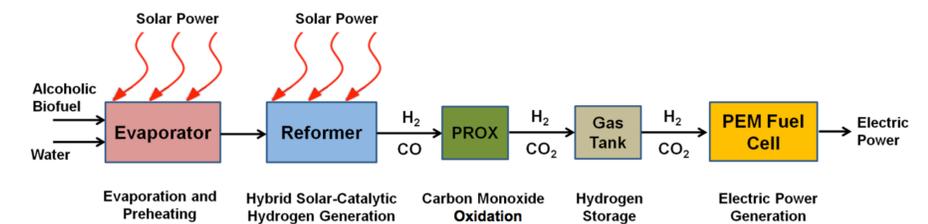
- Left graph shows fluid outlet temperature as a function of concentration ratio
 - Temperature increase starts to diminish as concentration ratio increases
- Right graph shows typical outlet temperatures expected over one day
 - Useful to determine when to operate collector; assumes ideal conditions



- Left graph shows collector efficiency as a function of mass flow rate
 - Middle region ideal: preserves efficiency while reaching high temperatures
- Right graph shows temperature profile as a function of position
 - Useful to determine size of collector required to achieve temperatures

Conclusions

- Concentrated solar power has the potential to increase fluid temperatures to values necessary to reform heavier biofuels into hydrogen
- Accuracy of model confirmed within ~10% when compared with historical experimental data on modeled non-concentrated system
 - Current model over-predicts outlet temperature at stagnation
 - Due to heat losses not accounted for in model (conductive losses, etc.)
- Steam reforming of heavier biofuels requires temperatures on the order of 1000 K, implying concentration ratio of 20 is required



Next Steps

- Refine the model:
 - Incorporate chemical reaction of steam reforming
 - Endothermic reaction with catalyst acts as heat sink, so model currently over-predicts temperatures of all parts of the collector
 - Model water as the working fluid and determine temperature increase
- Manufacture new absorption coating:
 - Current absorption coating degrades in high temperature applications
 - Testing of collector requires fabrication of new solar absorption coating
 - Must be able to both withstand high temperatures and perform efficiently
- Experimentally verify numerical predictions:
 - Determine best method of concentrating solar irradiance on collector
 - Use previously developed experimental methods to measure temperatures inside collector and at outlet versus mass flowrates and concentration

References

[1] Real et al. "Novel non-concentrating solar collector for intermediate-temperature energy capture." T-SEL MEMS Duke University. *Solar Energy*. Vol. 108 pp 421-431. Oct. 2014

[2] Gu et al. "Theoretical analysis of a novel, portable, CPC-based solar thermal collector for methanol reforming." School of Mechanical and Manufacturing Engineering, University of New South Wales. *Applied Energy*. Vol. 119 pp. 467-475. Apr. 2014

[3] Xu et al. "Methane Steam Reforming, Methanation and Water-Gas Shift: 1. Intrinsic Kinetics." *AIChE Journal*. Vol. 35, No. 1. pp. 88 – 96. Jan. 1989